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CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl 2 does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl 2 is an ionic solid. In the crystal form, its ions are locked in position. Dissolving

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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Match the four compounds on the right to their descriptions on the left. b an ionic compound that is quite soluble in water (a) HCl c an ionic compound that is not very soluble in water (b) NaNO 3

13 Ions in Aqueous Solutions and Colligative Properties

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